

Use of Cyclodextrins in ESR-Spin Trapping Experiments of Superoxide

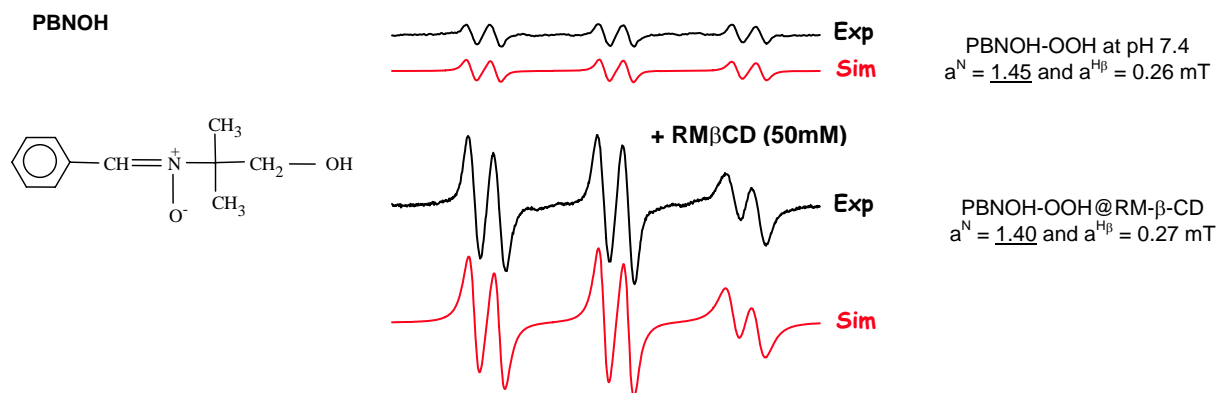
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Cyclodextrins are versatile biocompatible macrocycles based on 1,4-glucopyranose units with a truncated cone shape, which present outside hydrophilic sites and an inside hydrophobic cavity allowing the complexation of a wide variety of compounds in aqueous solution including stable nitroxides.¹

We used DMPO, DEPMPO and PBN to trap superoxide in the presence of methylated β -cyclodextrins.² The presence of β -cyclodextrins greatly stabilizes the PBN-superoxide spin adduct which is usually hardly detected.^{2b} The same phenomenon was observed with PBN analogues, DEPMPO,^{2a} EMPO and DMPO-superoxide spin adducts. PBN- and DEPMPO-superoxide adducts were partially protected against L-ascorbate monoanion reduction.^{2,3} In order to design the best candidate (spin trap – cyclodextrin) for our ESR spin trapping experiments, we studied the parameters concerning the associations' cyclodextrins-spin traps and cyclodextrins-nitroxides. The interactions between the simple linear traps (nitrones) and DM- β -CD were studied by NMR techniques to determine the stoichiometry. Binding constants were further evaluated by using ¹H NMR titrations. Secondly, ESR-spin trapping titrations in the presence of increasing cyclodextrin concentrations provided specifically spectral changes. Data from ESR titrations were analyzed by a novel two dimensional-ESR simulation program to afford stoichiometry and binding constants of the spin adduct@cyclodextrin systems. All the studied spin traps as well as spin adducts form inclusion complexes with DM- and RM- β -CD. All the spin trap@cyclodextrin complexes presented a 1:1 stoichiometry. Most of all the nitroxides could form 1:1 and 1:2 complexes depending on the nitroxide structure and the cyclodextrin concentration. Superoxide spin adduct ESR signal intensity depended on the nitrone-cyclodextrin association. Based on the different results we got, we studied the complexation of various stable nitroxides with different β -CD and prepared covalent nitroxide and nitrone- β -CD.



¹ Y. Kotake, and E.G. Janzen, *J. Am. Chem. Soc.*, 1992, **114**, 2872. G. Gagnaire, J. Michon, and J.L. Pierre, *New J. Chem.*, 1992, **16**, 915 ; Y. Kotake, and E.G. Janzen, *J. Am. Chem. Soc.*, 1989, **111**, 7319. A. Jeunet, B. Nickel, and A. Rassat, *New J. Chem.*, 1986, **10**, 123 ; J. Martinie, J. Michon, and A. Rassat, *J. Am. Chem. Soc.*, 1975, **97**, 1818 ; R.M. Patton, and E.T. Kaiser, *J. Am. Chem. Soc.*, 1970, **92**, 4723

² a. H. Karoui, A. Rockenbauer, S. Pietri, P. Tordo, *Chem. Comm.*, 2002, **24**, 3030-3031; b. H. Karoui, P. Tordo, *Tetrahedron Lett.*, 2004, **45**, 1043-1045.

³ D. Bardelang, A. Rockenbauer, H. Karoui, J.-P. Finet, P. Tordo. *J. Phys. Chem. B.*, 2005, **109**, 10521-10530.